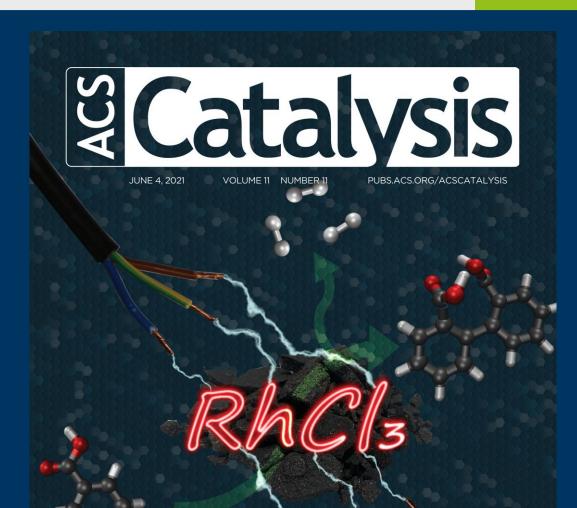
RUHR-UNIVERSITÄT BOCHUM

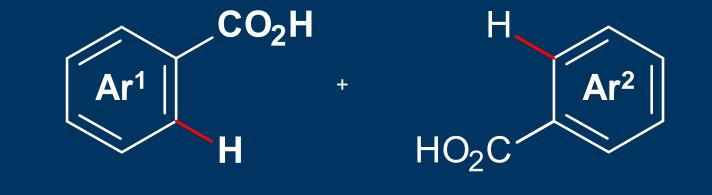
2,2'-Biaryldicarboxylate Synthesis via Electrocatalytic **Dehydrogenative C–H/C–H Coupling of Benzoic Acids**

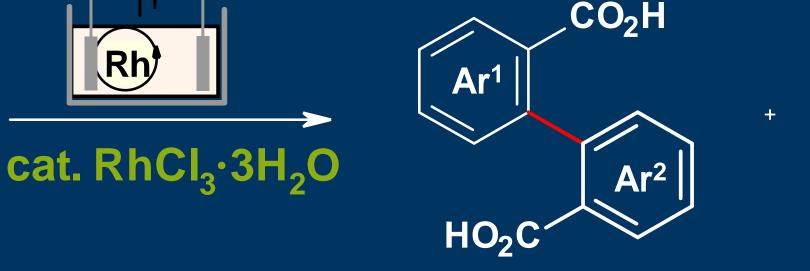
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2,2'-Biaryldicarboxylates are important functionalities in bioactive compounds, functional materials, and chiral catalysts. These compounds have been found to be conveniently accessible from benzoic acids via Rh-catalyzed electrooxidative C–H/C–H couplings, giving valuable dihydrogen as the byproduct. In an undivided cell with Pt electrodes, RhCl₃·3H₂O catalyzes the oxidative carboxylate-directed ortho-homocoupling of various aromatic acids with a current efficiency of 67%. The protocol is operationally simple, tolerates a wide variety of functional groups, and does not require the exclusion of air and moisture. Heterodimerizations via cross-dehydrogenative couplings of naphthyl-1-carboxylic acids with acrylic or benzoic acids have been proven to work as well.





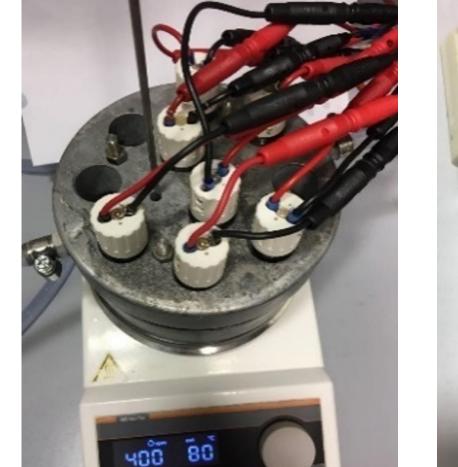






Introduction

In chiral catalysis, bioactive compounds or functional materials, 2,2'biaryldicarboxylates are often required as functional building blocks. However, conditions of biaryl synthesis usually require multistep prefunctionalization or wasteful oxidants.^[1] The use of electricity as an inexpensive and waste-free oxidant opens new opportunities for the development of sustainable C–H biaryl syntheses.^[2–4]



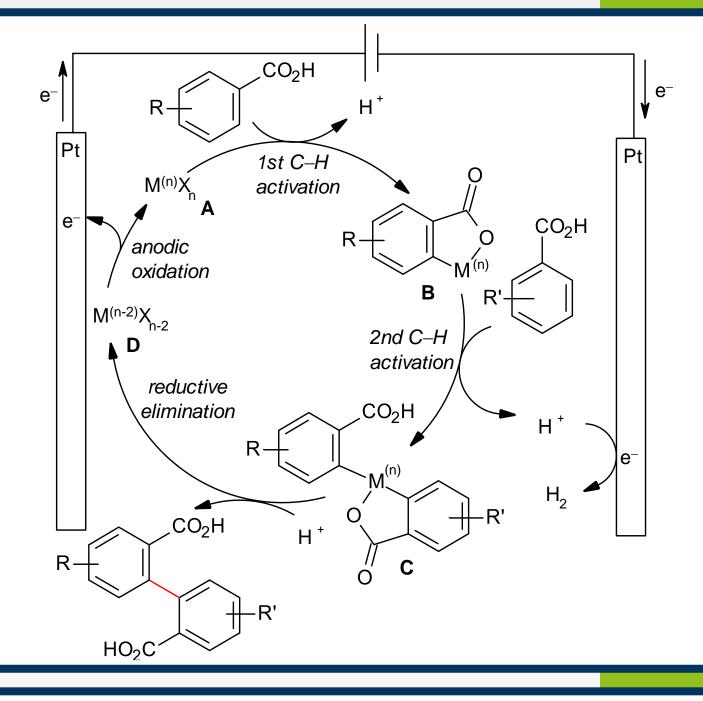
A) Chemical-oxidant-enabled C–H/C–H coupling H₂O + M B) Electrooxidative C–H/C–H homocoupling assisted by pyridine cathode 10 mol% [Pd] H_2SO_4 (aq.) Ar , CH₂CN + H₂ divided cell Pt(+)||Pt(-), 20 mA, 90 °C C) This work: Electrooxidative C-H/C-H coupling assisted by carboxylate CO₂H CO₂H

Ar¹

Proposed Mechanism

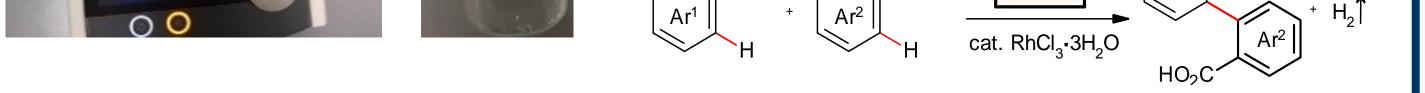
H₂

- C–H /C–H activation
- Rhodium catalyst
- electricity as "green" oxidant
- valuable H₂ as byproduct

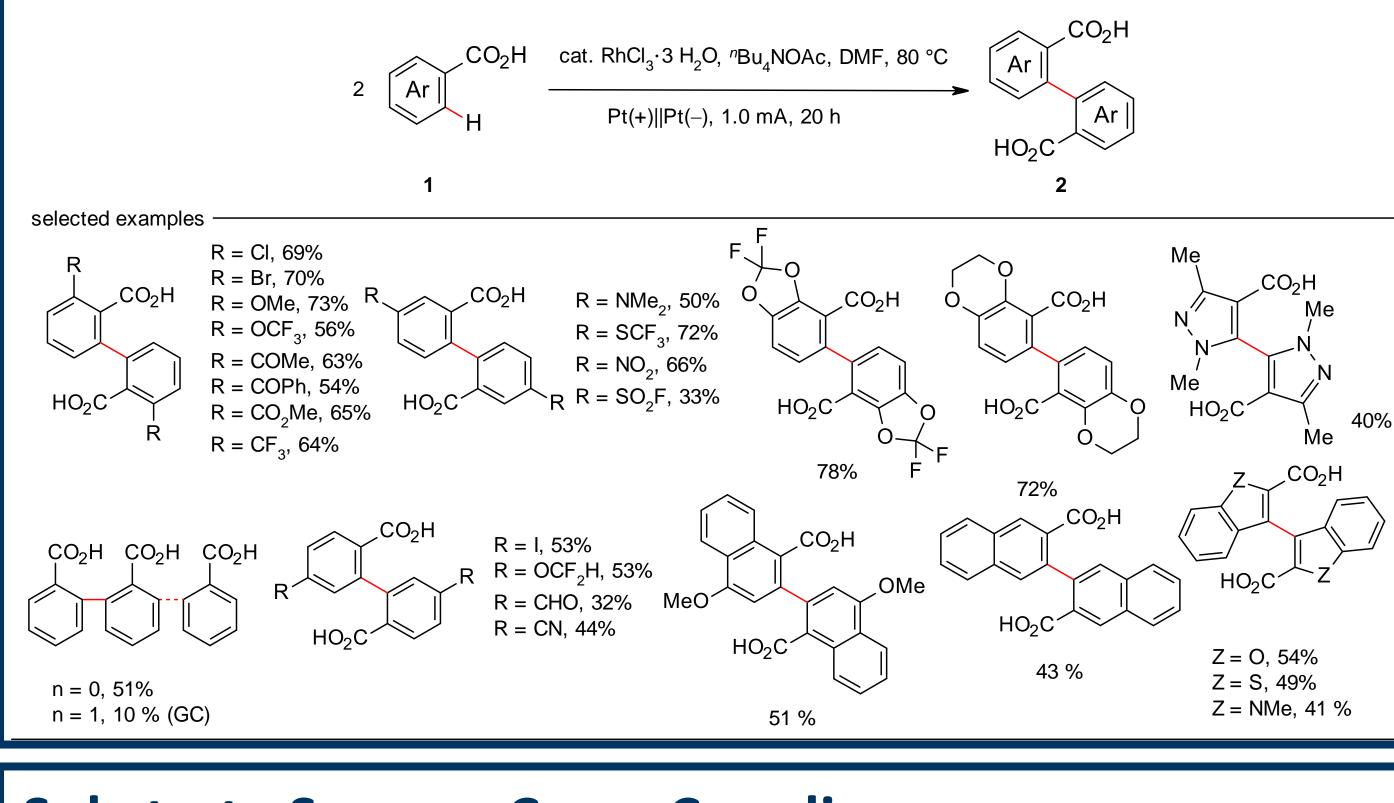


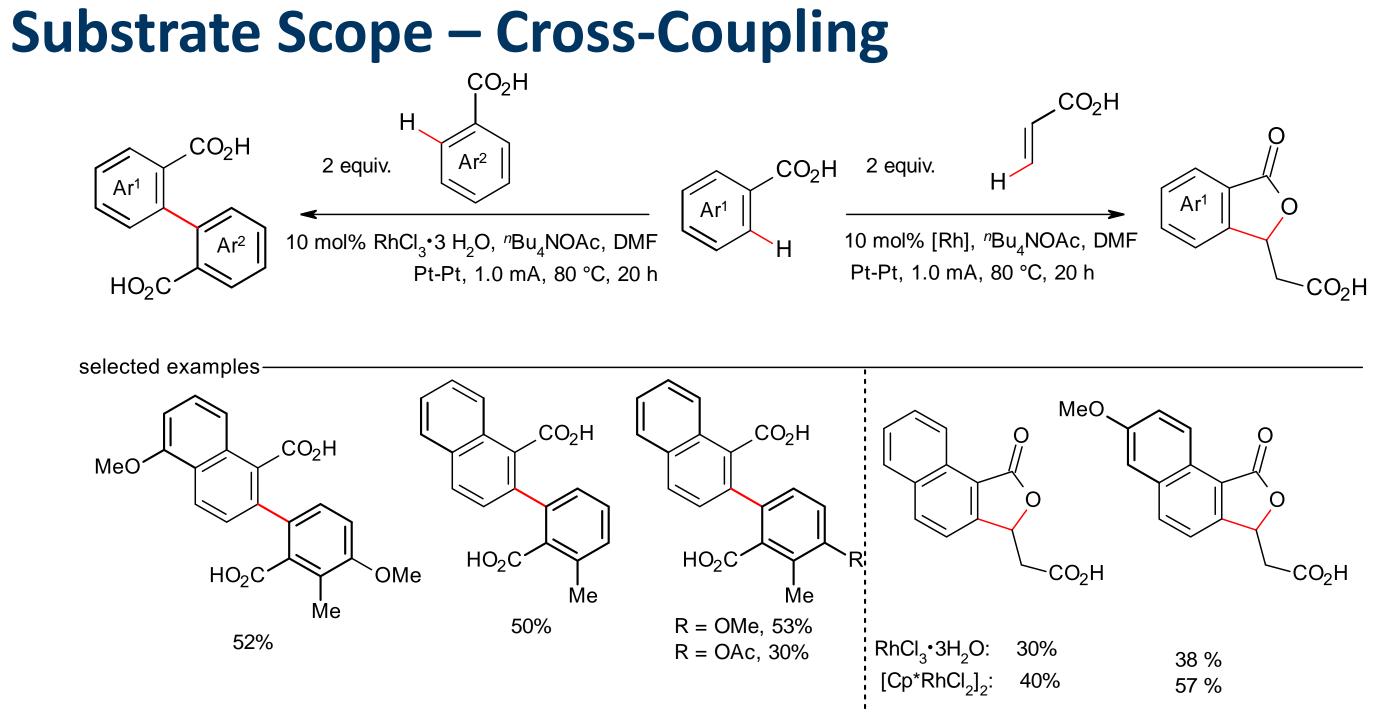
Mechanistic Investigations

rapid & reversable C–H metalation step



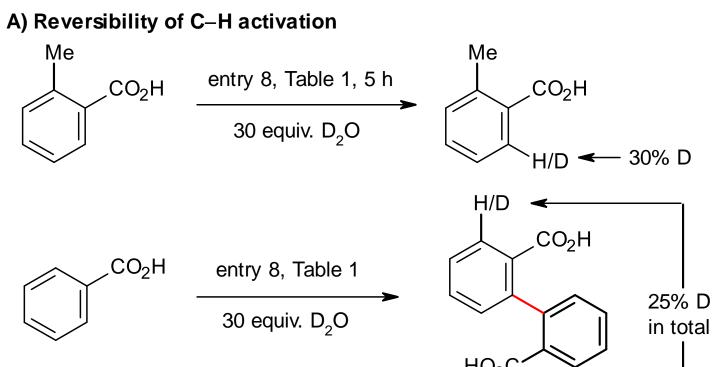
Substrate Scope – Homocoupling





kinetic isotope effect

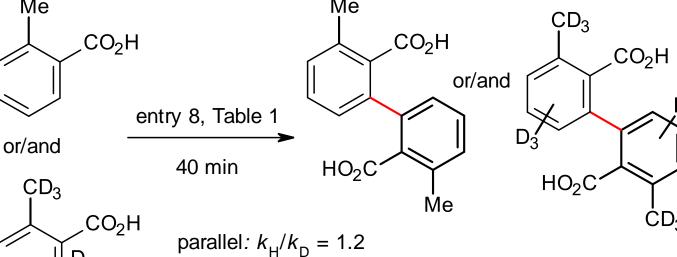
regeneration of Rh^{III} is rate-determining

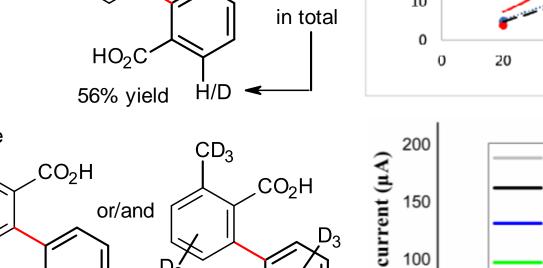


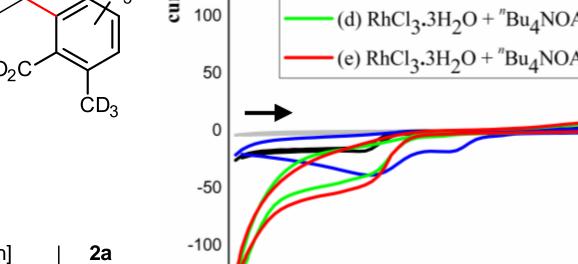


competitive: $k_{\rm H}/k_{\rm D} = 1.0$

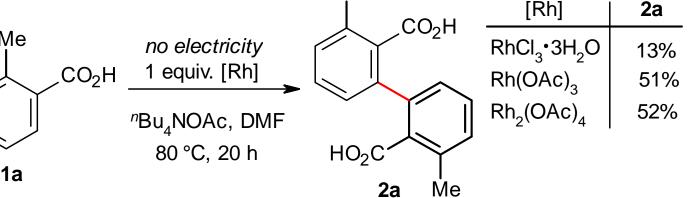
C) Reductive elimination and oxidation

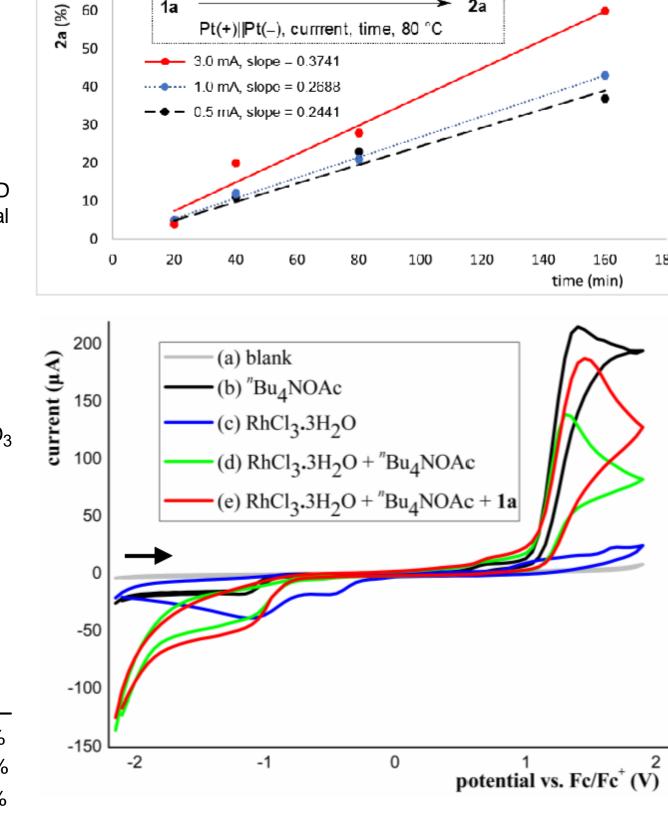






D) Correlation of current and reaction rate





7 mol% RhCl₃·3H₂O

Pt(+)||Pt(-), currrent, time, 80 °C

References

[1] J. F. Goebel, Z. Zeng, L. J. Gooßen, Synthesis 2022, 54, 565. [2] Z. Zeng, J. F. Goebel, X. Liu, L. J. Gooßen, ACS Catal. 2021, 11, 6626.

[3] C.-J. Li, Acc. Chem. Res. 2009, 42, 335.

[4] N. Sauermann, T. H. Meyer, Y. Qiu, L. Ackermann, ACS Catal. 2018, 8, 7086.

Conclusion

In conclusion, the newly developed electrocatalytic dehydrogenative C–H/C–H coupling allows for the formation of various biaryl dicarboxylates under mild conditions. The reaction is operationally simple and easily scalable, has an excellent functional group compatibility, and is orthogonal to common cross-coupling strategies.

We gratefully acknowledge financial support from:





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