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## **$[(\text{bmpyr})_2\{\text{Zn}(\text{OC}_6\text{H}_3(\text{NO}_2)_2)_4\}]$ : Influence of an Ionic Liquid on Liquid/Liquid Extraction of Metal Ions in a Biphasic System**

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**Abstract.** The compound  $[(\text{bmpyr})_2\{\text{Zn}(\text{OR})_4\}]$  (OR = 2,4-dinitrophenolate) has been prepared from  $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  and sodium 2,4-dinitrophenolate in a biphasic aqueous ionic liquid (Butyl-methyl-pyrrolidinium trifluoromethylsulfonate  $[\text{bmpyr}][\text{OTf}]$ ) system. The presence of the anionic zinc complex in  $[\text{bmpyr}][\text{OTf}]$  is made possible by the exchange of the ionic liquid anions into the aqueous phase for the zinc complex.  $[(\text{bmpyr})_2\{\text{Zn}(\text{OR})_4\}]$  was characteri-

zed in solution by  $^{13}\text{C}$ - and  $^1\text{H}$ -NMR spectroscopy and in the solid state by crystal structure determination. The zinc complex represents the first type of a zinc complex with more than two phenolate ligands.

**Keywords:** Ionic liquid; Biphasic system; 2,4-Dinitrophenolate; Zinc; Crystal structures